

Subscriber access provided by ISTANBUL TEKNIK UNIV

Isolation, Nmr Studies, and Biological Activities of Onopordopicrin from Centaurea sonchifolia

Greg Lonergan, Eugene Routsi, Theodoros Georgiadis, George Agelis, John Hondrelis, John Matsoukas, Linda K. Larsen, and Faith R. Caplan

> J. Nat. Prod., 1992, 55 (2), 225-228• DOI: 10.1021/np50080a012 • Publication Date (Web): 01 July 2004

Downloaded from http://pubs.acs.org on April 4, 2009

More About This Article

The permalink http://dx.doi.org/10.1021/np50080a012 provides access to:

- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article



Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

ISOLATION, NMR STUDIES, AND BIOLOGICAL ACTIVITIES OF ONOPORDOPICRIN FROM CENTAUREA SONCHIFOLIA

GREG LONERGAN,

Department of Chemistry, University of New Brunswick, Fredericton, New Brunswick, Canada

EUGENE ROUTSI, THEODOROS GEORGIADIS,

Department of Biology

George Agelis, John Hondrelis, John Matsoukas,*

Department of Chemistry, University of Patras, Patras, Greece

LINDA K. LARSEN, and FAITH R. CAPLAN

Department of Chemistry, University of Hawaii, Honolulu, Hawaii 96822

ABSTRACT.—A sesquiterpene lactone, onopordopicrin [1], has been isolated from Centaurea sonchifolia. Its structure was established by 2D nmr (${}^{1}H{}^{-1}H$ and ${}^{13}C{}^{-1}H$ correlations), and the conformation in CHCl₃ was examined by nOe studies. Cytotoxic, antibacterial, and antifungal activities are reported.

Onopordopicrin [1] is known to be major constituent of the genus Onopordone (1-4), with occasional appearance in some species of other genera (5). Previous investigation of the aerial parts of Centaurea sonchifolia L. from the Canary Islands resulted in the isolation of acetvlartemisiifolin (6) and artemisiifolin(7). Investigation of the aerial parts of C. sonchifolia from locations in southwestern Greece, however, resulted in the isolation of onopordopicrin, which was found to be the principal constituent of the lactone mixture. The structure was determined by ¹³C-¹H correlation experiments, COSY, DEPT, and other 2D nmr evidence. The conformation in CDCl₃ solution was studied by nOe enhancement spectroscopy. This lactone

shows cytotoxic activity against the KB cell line (a human epidermoid carcinoma of the nasopharynx). It also inhibits the growth of *Staphylococcus aureus*.

Table 1 reports the 200 MHz ¹H-nmr spectral assignments of onopordopicrin [1] in CDCl₃, including assignments of all proton resonances, ¹H chemical shifts, multiplicities, and proton-proton correlations derived from a COSY experiment. Table 2 reports the ¹³C spectrum of onopordopicrin in CDCl₃ including assignments of all carbon resonances, ¹³C chemical shifts, multiplicities as derived from DEPT experiments, and C-H correlations as derived from a C-H heteronuclear correlation experiment.

NOe enhancement experiments (8) of 1 (Table 3) indicated the α position of



internar standard).					
Proton	δ(ppm)	multiplicity	COSY		
H-1	5.02	dd (br)	2		
H-2	2.24	m	1		
H-2'	2.00	m	1		
Η-3α	2.24	m	3β, 2β		
Η-3β	2.62	m	3α, 2, 2'		
H-5	4.85	d	15, 15', 6		
Η-6β	5.14	dd	5,7α		
Η-7α	3.11	m	6, 13′		
Η-8β	5.19	dd (br)	9α		
Η-9α	2.62	m	8β		
Η-9β	2.00	m	_		
H-13	6.29	d	7α		
H-13'	5.81	d	7α		
H-14	1.51	s (br)	9α		
H-15	4.23	d	5		
H-15'	4.07	d	5		
H-18	4.31	s (br)	19, 19′, 18′		
H-19	6.30	s (br)	19′, 18, 18′		
H-19'	6.00	s (br)	18, 18', 19		

 TABLE 1.
 ¹H-nmr Data of Onopordopicrin

 [1] (200 MHz, in CDCl₃, TMS as internal standard).

H-1, H-5, and H-7. Thus, saturation of H-7 at δ 3.11 ppm resulted in enhancement of H-5 at δ 4.85 ppm (10%). Similarly, saturation of H-5 resulted in

 TABLE 2.
 13C-nmr Data of Onopordopicrin

 [1] (200 MHz, CDCl₃, TMS as internal standard).

Carbon	δ(ppm)	multiplicity	Proton correlation
C-1	129.67	d	5.02
C-2	26.04	t	2.24/2.00
C-3	34.49	t	2.24/2.62
C-4	144.50	S	_
C-5	127.86	d	4.85
C-6	77.28	d	5.14
C-7	52.83	d	3.11
C-8	72.85	d	5.19
C-9	48.80	t	2.62/2.00
C-10	139.60	s	_
C-11	132.10	s	
C-12	165.05	S	—
C-13	125.56	t	6.29/5.81
C-14	16.55	g	1.51
C-15	60.48	t	4.23/4.07
C-16	170.49	s	—
C-17	135.24	S	—
C-18	60.91	t	4.31
C-19	125.67	t	6.00/6.30

 TABLE 3.
 NOe Difference Spectral Data of Onopordopicrin [1].

enhancement of H-7 and H-1 at δ 3.11 and δ 5.02 ppm, supporting the α position assignment of H-1, H-5, and H-7. NOe experiments also indicated proximity and β location of H-6 and H-8. Thus, saturation of the axial Me-14 resonance at δ 1.51 ppm resulted in enhancement of H-6 and H-8 at δ 5.14 ppm and δ 5.19 ppm, respectively. Similarly, saturation of H-8 at δ 5.19 ppm produced enhancement of the Me-14 and H-6 resonances at δ_1 1.51 ppm and δ_2 5.14 ppm, respectively. No enhancement was observed between H-6 and H-7, thus supporting a trans-fused lactone ring structure.

A strong nOe was observed between H-13 and H-13' (at $\delta_1 6.29$ ppm and δ_2 5.81 ppm) with H-19 and H-19' at δ_1 6.30 ppm and δ_2 6.00 ppm, indicating close proximity of the two double bonds (C-11-C-13 and C-17-C-19). Thus, irradiation at H-13 resulted in enhancement of H-19 and H-19'. Similar enhancement was observed for H-19 and H-19' upon irradiation of H-13'. The close proximity of the two methylene groups, CH2-19 and CH2-13, and of the two double bonds, C-11-C-13 and C-17–C-19, may be the result of π - π interactions. This interaction is reminiscent of aromatic ring stacking observed in biological systems (9, 10), and may be important for bioactivity.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.-1D and 2D ¹H-nmr spectra were recorded on a Varian XL-200 instrument at 200 MHz in CDCl₃, and ¹³C-nmr spectra were recorded at 50 MHz in CDCl₃. The COSY spectrum was recorded using standard Varian software (Vs 6.2, Rev. B) with spectral widths of 1500 Hz in both frequency dimensions. Sixteen transients were recorded for each of 256 t1 increments, with a recycle delay time of 3 sec per transient for a total experiment time of 4 h. One order of zero-filling was applied in t₁ prior to Fourier transformation to produce a digital resolution of 0.015 ppm. Chemical shifts are reported in δ units (ppm) relative to TMS as internal standard. Ir spectra were recorded with a Perkin-Elmer 457 instrument, while ms were obtained with a gc-ms Finnigan 4021 EI-Cl-INCOS system at 70 eV. Purity of onopordopicrin was checked also by gc analysis performed on a Varian 6000 instrument equipped with a flame ionization detector.

PLANT MATERIAL.—The plant material was collected in Kalogria beach, near Patras, Greece, in March 1988, and identified by Professor Th. Georgiadis, Department of Plant Biology, Division of Plant Taxonomy, University of Patras. A voucher specimen, No. 3710, is deposited at the Herbarium of the University of Patras.

ISOLATION OF THE LACTONE.—Dried and ground aerial plant material (500 g) of *C. sonchifolia* was extracted with three 2-liter portions of MeOH. The MeOH solution was evaporated and concentrated to 100 ml H_2O (1000 ml) was then added, followed by addition of lead acetate (10 g) dissolved in 100 ml of H_2O . The mixture stayed at room temperature for 30 min and then was filtered. The filtrate was extracted with CHCl₃ (1000 ml), and the CHCl₃ solution was dried over MgSO₄ and evaporated to give the lactone fraction as a crude syrup (10 g).

A portion (500 mg) of the lactone fraction was chromatographed on three preparative tlc plates $(20 \times 20 \text{ cm}, 2 \text{ mm})$ to afford the major lactone in oily form (177 mg). The lactone was extracted from the preparative Si gel plate with 200 ml of CHCl₃-MeOH (9:1) and was found to be homogeneous on tlc systems of CHCl₃-MeOH (9:1) and CHCl₃-Me₂CO (3:2).

The oily lactone was subjected to gc-ms analysis, which confirmed its purity. It was, furthermore, subjected to 2D nmr and distortionless enhancement by polarization transfer (DEPT) experiments (Table 1–3). The lactone was crystallized from CHCl₃/Et₂O and identified by comparison with authentic onopordopicrin: mp 55– 58°; $[\alpha]^{25}D + 16.2^{\circ}$ (c = 0.5, MeOH). The eims, ir, and ¹H- and ¹³C-nmr spectra were also identical with those of the reference: ms m/z (%) 85 (100), 91 (34), 119 (52), 147 (22), 215 (8), 228 (5), 246 (3); ir (KBr) 3400 (hydroxyl), 2920 (methylene), 1750 (γ lactone), 1700 cm⁻¹ (ester carbonyl unsaturated).

BIOLOGICAL ACTIVITIES.—In vitro cytotoxic activity of onopordopicrin, using the human epidermoid carcinoma cell line KB (ATCC CCL 17), was determined by methods previously described (11,12). The ED₅₀ value was 0.85 $\mu g/$ ml. Cytotoxicity and antimicrobial activity of α methylene- γ -lactone-bearing sesquiterpene lactones has been earlier reported (13–15).

Antibacterial activity experiments were conducted applying the disk diffusion assay (16). Onopordopicrin was found to possess antibacterial activity against *St. aureus* (ATCC 25923) but not against other Gram-positive and Gram-negative bacteria that were tested.

No antifungal activity at 250 μ g/disk was observed against the test organisms, *Candida albicans* (strain A-26, Eli Lilly), *Trichophyton mentagrophytes* (strain A-23, Eli Lilly), and *Saccharomyces cerevisiae* (strain M25, generously provided by Dr. Bruce Adams, Department of Microbiology, University of Hawaii).

ACKNOWLEDGMENTS

This work was supported by the Greek Ministry of Research and Technology. We are grateful to Prof. P. Scheuer of the University of Hawaii at Manoa and Prof. J. Findlay of the University of New Brunswick, Canada, for providing facilities for part of this work. We thank Dr. Richard Moore and Dr. Greg Patterson, University of Hawaii at Manoa, for the cytotoxicity and antifungal assays. We also thank Dr. Larry Calhoun, University of New Brunswick, for the determination of nmr spectra and Prof. B. Drożdż, Medical Academy, Poznan, Poland, for providing us with an authentic sample of onopordopicrin.

LITERATURE CITED

- B. Drożdż, M. Holub, Z. Samek, V. Herout, and F. Sorm, Collect. Czech. Chem. Commun., 33, 1730 (1968).
- S.M. Kfafagy, A.M. Metwall, and A.A. Omar, *Pharmazie*, **32**, 123 (1977).
- A. Rustaiyan, B. Ahmadi, J. Jakypovic, and F. Bohlmann, *Phytochemistry*, 25, 1659 (1986).
- 4. G. Nowak, B. Drożdź, and T. Georgiadis, Acta Soc. Bot. Pol., 53, 199 (1984).
- A.G. Gonzalez, J.M. Arteaga, and J.L. Breton, *Phytochemistry*, **12**, 2997 (1983).
- T.H. Porter, T.J. Mabry, H. Yoshioka, and N.H. Fischer, *Phytochemistry*, 9, 199 (1970).
- A.G. Gonzalez, J.B. Barrera, T.Z. Garcia, and F.E. Rosas, *Phytochemistry*, 23, 2071 (1984).

- J.H. Noggle and R.E. Shirmer, "The Nuclear Overhauser Effect," Academic Press, New York, 1971, pp. 167-243.
- 9. J.M. Matsoukas and G.J. Moore, Biochem. Biophys. Res. Commun., 123, 434 (1984).
- 10. J.M. Matsoukas and D. Theodoropoulos, Org. Magn. Reson., 12, 393 (1977).
- J.B. Stewart, V. Bornemmann, J.V. Chen, R.E. Moore, F.R. Caplan, H. Caruso, L.K. Larsen, and G.M.L. Patterson, J. Antibiot., 41, 1048 (1988).
- R.I. Geran, N.H. Greenberg, M.M. McDonald, A.M. Schumacher, and B.J.

Abbott, Cancer Chemother. Rep., 3, 1 (1972).

- K.H. Lee, E.S. Huang, C. Piantadosi, J.S. Pagano, and T.A. Geissman, *Cancer Res.*, **31**, 1649 (1971).
- 14. S.M. Kupchan, M.A. Eakin, and A.M. Thomas, J. Med. Chem., 14, 1147 (1971).
- K.H. Lee, T. Ibuka, R.Y. Wu, and T.A. Geissman, *Phytochemistry*, 16, 1177 (1977).
- 16. J.A. Findlay and A.D. Patil, *Phytochemistry*, 25, 548 (1986).

Received 29 October 1990